



Impact Of Plasma Bubbling On Cow Milk: Microbial Reduction And Improvement Of Milk Quality

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Abstract: Raw milk has high nutritive value and is used as an important dietary supplement for humans. But, microbial contamination in milk has been a problem for era, while existing thermal technology often deteriorates the quality of milk and its valuable ingredients. Again, consumer demand for least processing technology without degradation of the product quality forced the scientist to develop novel methods relevant to the application of non-thermal technology which does not create any detrimental effect to the composition of milk are now being under scrutiny. One of the novel methods is plasma bubbling technique and is not yet examined for liquid foods like milk. In the present study, the plasma bubbling system was established for the decontamination of raw cow milk. The plasma bubbling was generated at voltage of 160V, for 5, 10 and 15 minutes (min) and was evaluated for microbial reduction at an air flow rate of 5 and 10 Litre/hour (L/h). It accounted for a maximum 1.33 log reduction for coliforms at 160V, flow rate of air was 10 L/h, for a 100mL of milk sample with 15 min exposure time, while for yeast the log reduction was 1.40. The plasma bubbled milk was analysed for its quality evaluation such as pH, acidity, colour and lactose content of milk. The value of pH was found to be 6.77 at 160V, 10L/h, 15 min and 100mL of sample volume while the control value of pH was 6.60. The findings from this study revealed that the atmospheric plasma bubbling system could be used for the pasteurisation of raw cow milk by reducing the microbial load without compromising milk quality. This work on novel atmospheric plasma bubbling is an initiative for the pasteurization of raw cow milk, which could have a potential impact on the food industry in future.

Key Words: Plasma bubbling, Safety, Quality, Milk

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1. INTRODUCTION

Milk is a very good source of carbohydrates, fatty acids, high-quality protein content and various micronutrients, such as vitamins and minerals¹. It was reported that by 2025, global per capita dairy consumption will increase to 12.5%². Besides, dairy products are considered as a high value of nutrition along with health-promoting foods for the prevention of several diseases such as sarcopenia, metabolic syndrome, osteoporosis, cardiovascular disease, cognitive decline and digestive disorders³. But before consumption, the milk should be subjected to thermal processing which improves microbial safety but they extensively damage the nutritional and physicochemical properties⁴⁻⁶. However, recent interest in the consumption of raw milk has led to the scientist for consideration of alternative technologies that will not compromise the quality and safety of milk i.e., nonthermal technique, which can meet microbial food safety as well as enhance the physical, nutritional, sensory characteristics of the products and helps to preserve the unstable bioactive compounds^{7,8}. The generation of non-thermal (cold plasma) plasma occurs at the non-equilibrium condition when the temperature of the electron is higher than the heavy particles⁹. Plasma contains a mixture of particles, molecules, ions and reactive oxygen species which acts simultaneously for the decontamination of the target¹⁰. This characteristic of plasma has led researchers to evaluate different cold plasma generation devices which can be adjusted accordingly to maintain a suitable condition for different applications. Cold plasma application in the food sector includes sterilization, decontamination, pesticide reduction, property modification and germination¹¹. Cold plasma has shown decontaminating properties for various materials such as poultry, milk, water, living cells, meat, fresh fruit and vegetables, because of its capacity to kill microorganisms, including bacteria, yeasts, fungi and algae. It has become a well-known technology in the field of the food industry¹²⁻¹⁷. The key characteristics for the decontamination of microorganisms are the UV light, source

of free radicals, ions, electrons generated during cold plasma treatment. Radicals like reactive oxygen species (ROS) can directly decontaminate microbes by gaseous phase¹⁸. However, in the case of liquid foods, the effect of plasma is complex because of its free-flowing nature¹⁹. Several studies have been conducted in recent years on liquid food for the decontamination purpose as well as for the quality evolution. Thus, the demand is increasing for establishing a novel approach for the pasteurization of liquid foods. The various non-thermal approaches were conducted for milk quality evolution such as ultraviolet light²⁰, ultrasound²¹, cold plasma²² and pulsed electric field²³ which have been studied for the replacement of the thermal pasteurization. Previously, several studies have focused on the decontamination of milk by using cold plasma technology^{9,17-24}. Despite many studies conducted on decontaminating the capacity of cold plasma, there is limited scrutiny for the effect of the fourth state of matter (cold plasma) on the food products. The aim of the study is the use of cold atmospheric plasma gas, which has been bubbled through raw cow milk at different input flow rate of air and time with a constant voltage and to investigate the effect of input parameters on microbial and physicochemical properties of raw cow milk such as pH, titratable acidity, total soluble solids, colour.

2. MATERIALS AND METHODS

2.1 Experimental Setup

Cold plasma bubbling system was developed indigenously and was used with flow rates of air at 5 and 10 litres per hour (L/h) while using atmospheric air as feed gas²⁵. Plasma was generated at voltage 160V and was bubbled through 50, 100 and 150 mL of raw cow milk for 5, 10 and 15 min. The overall experimental set up was depicted in (Fig 1). Previously observed parameters for cold plasma were established with slight modification²⁵.

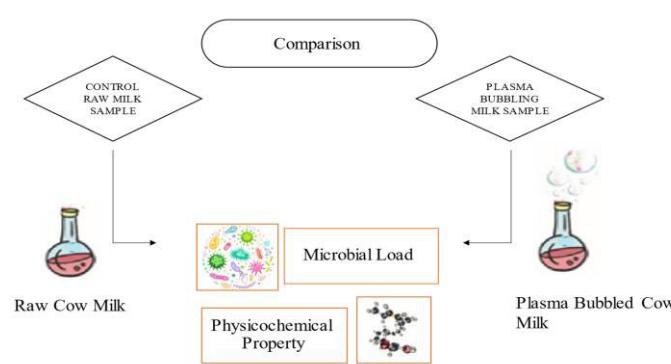


Fig 1. Schematic diagram of the overall experimental design

2.2 Microbial Analysis

In the present study, microbes analysed for milk samples were coliform, yeast²⁶ and was observed by using the spread plate method in triplicates²⁷. The control and treated cow raw milk samples were tested for coliform using violet red bile agar²⁸. For yeast, chloramphenicol yeast glucose agar was

used²⁹. The experimental analysis was performed in triplicate and the result was represented as log CFU/mL²⁶.

2.3 pH analysis

The pH of the raw cow milk and plasma bubbled cow milk sample was analysed by using the pH meter (Laqua PH1100, Horiba Scientific, Singapore) at ambient temperature³⁰.

2.4 Colour measurement

Plasma bubbled cow milk and fresh raw cow milk was poured into a cuvette cell (64 mm) and the colour of the cow milk was evaluated by colourimeter (Colour Flex EZ System, 45/0 LAV); the colour values L^* (lightness), a^* (redness) and b^* (yellowness) were determined³¹.

2.5 Titratable acidity

The titratable acidity was analysed by using 5 drops of 1% w/v phenolphthalein indicator and were added to the conical flask containing 20 mL of cow milk. The mixture was titrated using 0.1 N NaOH and the % of titratable acidity was calculated³⁰.

2.6 Total soluble solids

The TSS content was measured at room temperature by a digital refractometer (Erma, Japan, 0-80° Brix)³².

3. STATISTICAL ANALYSIS

SPSS version 22 statistical software (SPSS, Inc., United States) was used to analyse the results. The replication of all the experimental analyses was three times. The statistical analysis was done by using a one-way analysis of variance (ANOVA). The significant differences among the mean values were

determined by performing Duncan's multiple comparison tests at a confidence level of $p < 0.05$.

4. RESULTS AND DISCUSSION

4.1 Microbial Analysis

In this study, it was observed that the microbial reduction is based upon the processed parameters such as the input time, volume and flow rate of air. A log reduction of 1.33 was observed at (160 V), with a flow rate of air (10 L/h), 100 mL of the volume of cow milk and 15 min treated time in the plasma bubbling system. While for yeast 1.4 log reduction was observed for 160V, 10L/h, 100mL and 15 min treated parameters. This could be due to the generation of hydroxyl radicals HO^{\cdot} during cold plasma, which damages the cell wall³³. Non-thermal plasma produces gas such as ozone, nitric oxide, oxygen, hydroxyl radicals and could able to directly interact with the bacterial membrane¹⁴. Another study revealed that ROS and RNS plays a vital role in microbial cell death by breaking the double bond of the lipid bilayer present on the microbial cells, thus strong oxidative stress occurred which damages the transportation of macromolecules to the cell, therefore pathogen inactivation was observed³⁴. Again, it was observed that after exposure to cold plasma, different types of stress could be responsible for the death of yeast cell³⁵. A comparison of the microbial load was depicted (Fig 2).

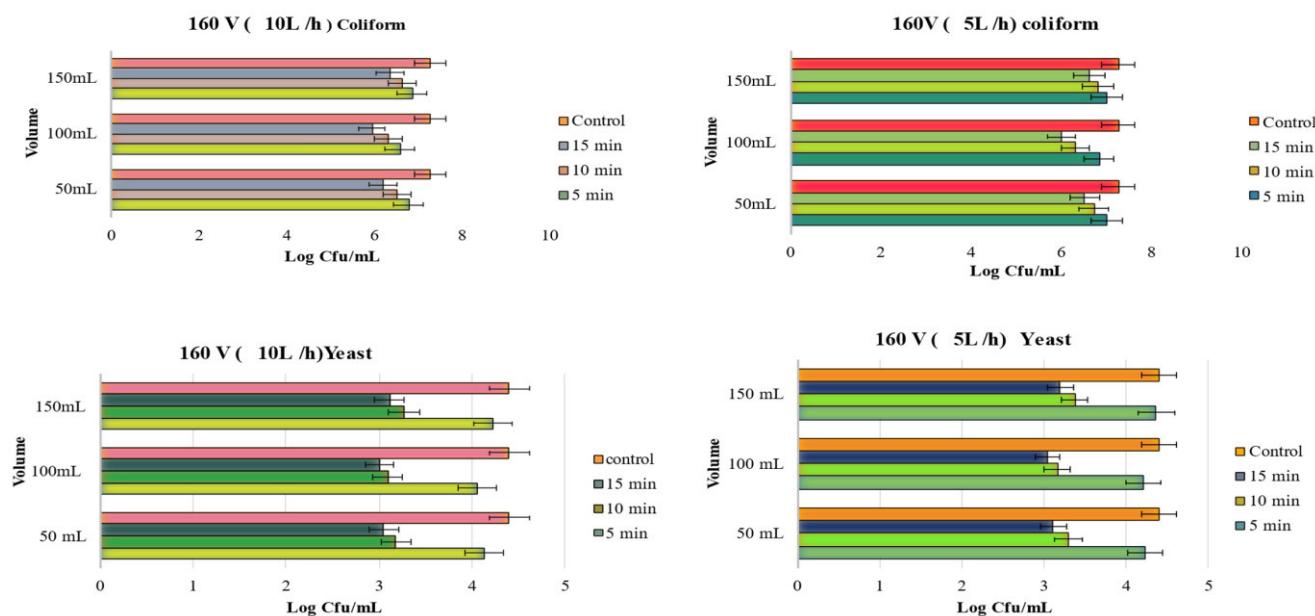


Fig 2. Reduction of microbial cell viability (Log CfU /mL) control and treated sample.

4.2 pH

The concentration of hydrogen is crucial in cow milk. In this study, the value of pH is slightly increasing in accordance with particular volume, high flow rate of air and time interval (Table 3). In the control sample, the pH value was 6.60 ± 0.011 while in 160V, 10L/h, 100mL, and 15 min treatment time the pH value was 6.77 ± 0.018 . The volume, flow rate and time play an important role in the increase of pH value. Again, previously it was reported that HO^{\cdot} radicals are produced during non-thermal treatment which could be responsible for the pH³⁶.

4.3 Colour

A difference in colour value was observed after being exposed to plasma bubbling treatment with a low volume of sample, high flow rate and higher treatment time. A significant decrease was observed for L^* , a^* and b^* value at 160V, 10L/h, 50mL, 15min processed parameters with respect to control colour value of milk (Table 2). A decrease in colour difference could be explained due to the partial degradation of pigment after exposure to cold plasma treatment³⁷⁻³⁸. Previously, it was observed that a slight colour

change has occurred in milk after exposure to cold plasma³⁹. In the present study, it was observed that the volume of the sample plays a major role in the colour difference in 50mL of

volume; the colour value decreases more than that of 100mL and 150mL

Table 2. Colour value of cow milk before and after plasma

Time (Minutes)	Voltage(V)	Flow Rate of air (L/h)	Volume (mL)	L*	a*	b*
Control						
5	160	5	50	88.76 ^e ±0.013	-0.91 ^s ±0.012	14.23 ^f ±0.013
			100	88.72 ^d ±0.005	-1.02 ^p ±0.012	14.21 ^f ±0.005
			150	88.74 ^e ±0.005	-0.97 ^q ±0.00	14.22 ^f ±0.006
			50	88.76 ^e ±0.012	-0.95 ^r ±0.01	14.23 ^f ±0.006
			100	88.70 ^d ±0.005	-1.15 ^m ±0.011	14.18 ^e ±0.012
		10	150	88.71 ^d ±0.006	-1.06 ⁿ ±0.005	14.20 ^e ±0.005
			50	88.74 ^e ±0.006	-1.00 ^o ±0.005	14.21 ^f ±0.005
			100	88.66 ^c ±0.00	-1.28 ^l ±0.006	14.15 ^d ±0.00
			150	88.69 ^d ±0.006	-1.12 ^k ±0.017	14.17 ^e ±0.005
			50	88.70 ^d ±0.00	-1.08 ^l ±0.00	14.19 ^e ±0.11
10	160	5	150	88.64 ^c ±0.017	-1.33 ^g ±0.023	14.12 ^d ±0.006
			50	88.67 ^c ±0.00	-1.21 ^h ±0.011	14.13 ^d ±0.005
			100	88.68 ^c ±0.006	-1.14 ⁱ ±0.00	14.16 ^e ±0.12
			150	88.57 ^a ±0.006	-1.59 ^d ±0.005	14.07 ^c ±0.011
			50	88.60 ^b ±0.023	-1.48 ^e ±0.00	14.10 ^d ±0.028
		10	100	88.61 ^b ±0.023	-1.30 ^f ±0.011	14.11 ^d ±0.012
			150	88.53 ^a ±0.017	-1.76 ^a ±0.005	14.02 ^a ±0.028
			50	88.56 ^a ±0.01	-1.64 ^b ±0.028	14.05 ^{ab} ±0.011
			100	88.59 ^b ±0.011	-1.53 ^c ±0.006	14.09 ^b ±0.012
			150			

The results were expressed as mean ±S.D. Mean values followed by different letters in the same row indicate significant differences ($p < 0.05$).

4.4 Titratable acidity

After exposure to plasma bubbling with a particular volume, high flow rate along with increasing in time interval, a gradual decrease in titratable acidity (TA) value of the milk was observed. The lowest value of titratable acidity was observed at 160V, 10 L/h, 100mL, 15 min i.e., 0.127 ± 0.015 % w/v lactic acid whereas the control sample value of lactic acid was 0.144 ± 0.012 w/v (Table 3). This could be explained due to an increase of hydroxyl HO[·] radicals formed by the decomposition of an additional water molecule during cold plasma⁴⁰.

4.5 Total soluble solids

In this study it was observed that with increasing flow rate and time the value of TSS was decreasing. A prominent decrease value of TSS was observed at 160V, 10 L/h, 100mL, 15 min i.e., 8.9° Brix; while the value for control was 10° Brix (Table 4). The decrease in TSS could be explained due to the formation of ozone during cold plasma generation which is responsible for de-polymerization of the macromolecule and thus responsible for the decrease in TSS value³³⁻⁴¹.

Table 3. Physicochemical Property of control and plasma bubbled milk sample.

Time (Min)	Voltage (V)	Flow rate of air (L/h)	Volume (mL)	pH	TA	TSS
Control						
5	160	5	50	6.60 ^a ±0.011	0.144 ^d ±0.012	10.2 ^d ±0.002
			100	6.61 ^a ±0.011	0.139 ^{cd} ±0.012	10.0 ^c ±0.00
			150	6.62 ^a ±0.015	0.138 ^c ±0.014	9.9 ^c ±0.003
			50	6.61 ^a ±0.02	0.139 ^{cd} ±0.021	10.1 ^{bc} ±0.013
			100	6.62 ^a ±0.021	0.137 ^c ±0.022	9.9 ^c ±0.014
		10	100	6.63 ^a ±0.012	0.136 ^c ±0.010	9.8 ^c ±0.011
			150	6.62 ^a ±0.010	0.137 ^c ±0.011	9.9 ^c ±0.006
			50	6.64 ^{ab} ±0.010	0.134 ^b ±0.011	9.8 ^c ±0.013
			100	6.65 ^b ±0.011	0.130 ^{ab} ±0.009	9.5 ^b ±0.005
			150	6.63 ^a ±0.011	0.135 ^{bc} ±0.009	9.8 ^c ±0.007
10	160	5	50	6.66 ^{bc} ±0.017	0.132 ^b ±0.018	9.6 ^b ±0.015
			100	6.70 ^c ±0.015	0.129 ^a ±0.012	9.4 ^b ±0.011
			150	6.65 ^b ±0.011	0.133 ^b ±0.010	9.7 ^b ±0.021
			50	6.71 ^c ±0.003	0.130 ^{ab} ±0.005	9.2 ^a ±0.005
			100	6.74 ^{cd} ±0.010	0.128 ^a ±0.011	9.1 ^a ±0.008
		10	150	6.70 ^c ±0.014	0.131 ^b ±0.013	9.3 ^a ±0.006
			50	6.72 ^c ±0.016	0.129 ^a ±0.015	9.0 ^a ±0.002
			100	6.77 ^d ±0.018	0.127 ^a ±0.015	8.9 ^a ±0.005
			150	6.71 ^c ±0.001	0.129 ^a ±0.002	9.1 ^a ±0.011

The results were expressed as mean \pm S.D. Mean values followed by different letters in the same row indicate significant differences ($p < 0.05$).

5. CONCLUSION

In the present study, plasma bubbling treatment of raw cow milk was done at a constant voltage (160V) and 1.33, 1.40 log reduction for coliforms and yeast was achieved at 160V, 100mL, 10L/h and 15min treatment respectively. While, atmospheric plasma bubbling technology maintain the physicochemical quality of milk with a non-detrimental effect. The plasma bubbling parameter at a voltage 160V, 10L/h flow rate of air, sample volume of 100mL and 15 min of time interval considered as the good parameter set up among the other set up parameters. However, the current study is a preliminary experiment for the plasma bubbling treatment for cow milk. Further optimization of process parameters along with milk quality and safety need to be examined. Again, the seasonal atmospheric gas needs to determine for better understanding the feed gas for cold plasma and its reaction with respect to the seasons.

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7. AUTHORS CONTRIBUTION STATEMENTS

Mrs Samarpita Dash conceived the practical work and wrote the manuscript. Dr R. Jaganmohan conceived the manuscript idea. All the authors discussed the methodology and results.

8. DECLARATION OF COMPETING INTEREST

The author declares there is no conflict of interest.

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